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### **Section Properties Of Water 2**

Section 2-2 Properties of Water(pages 40-43) This section describes the makeup of water molecules. It also explains what acidic solutions and basic

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solutions are. The Water Molecule(pages 40-41) 1. Is the following sentence true or false? A water molecule is neutral. 2. What results from the oxygen atom being at one end of a water molecule and the

### **Section 2-2 Properties of Water**

Start studying Section 2: Properties of

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## **Section 2: Properties of Water Flashcards | Quizlet**

negative (oxygen) atom a mixture that is the same throughout (homogeneous) substance present in greatest

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concentration; dissolves other substances. High specific heat—large amount of energy needed to produce an increase in temperature

Cohesion—water molecules “stick” to each other.

### **SECTION PROPERTIES OF WATER 2.2 Power Notes**

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- Hydrogen bonds are responsible for three important properties of water.
    - high specific heat
    - cohesion
    - adhesion
- 2.2 Properties of Water Many compounds dissolve in water. • A solution is formed when one substance dissolves in another.

## **2.2 Properties of Water**



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2-2 Properties of Water. Slide 5 of 44.  
The Water Molecule. Polarity However an oxygen atom, with 8 protons in its nucleus, has a much stronger attraction for electrons than does the hydrogen atom with a single proton.

## **2-2 Properties of Water**

-the attraction of water to molecules of

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other substances-The polar qualities of water will cause them to try to form bonds with various other molecules/substances-So for example, we see this in graduated cylinders - the water will curve due to the adhesive properties of water causing the formation of a meniscus.

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## **"Properties of Water" Chapter 2, Section 2 Flashcards ...**

Properties of Water. II. Properties of Water. 1. Ice and Liquid water structure 2. Cohesion / Surface Tension 3. High Heat Capacity 4. Solvent Properties 5. Dissociation: Acids & Bases / pH.

## **II. Properties of Water**

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heat capacity means the amount of heat energy required to increase its temperature. water's is relatively high because of water's multiple hydrogen bonds. this is good because large bodies of water are hardly affected. at a cellular level, water absorbs heat produced by cells, keeping it regulated.

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## **Biology Section 2.2 Properties of Water Flashcards | Quizlet**

Water can moderate temperature because of the two properties: high-specific heat and the high heat of vaporization. High-specific heat is the amount of energy that is absorbed or lost by one gram of a substance to change the temperature by 1 degree

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celsius. Water molecules form a lot of hydrogen bonds between one another.

## **5 Properties of Water - Owlcation - Education**

The absorption of electromagnetic radiation by water depends on the state of the water. The absorption in the gas phase occurs in three regions of the

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spectrum. Rotational transitions are responsible for absorption in the microwave and far-infrared, vibrational transitions in the mid-infrared and near-infrared. Vibrational bands have rotational fine structure. Electronic transitions occur in the vacuum ultraviolet regions. Liquid water has no rotational spectrum but does absorb in

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the microwave

## **Electromagnetic absorption by water - Wikipedia**

Water is the chemical substance with chemical formula  $H_2O$ ; one molecule of water has two hydrogen atoms covalently bonded to a single oxygen atom. Water is a tasteless, odorless



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liquid at ambient temperature and pressure. Liquid water has weak absorption bands at wavelengths of around 750 nm which cause it to appear to have a blue colour.

### **Properties of water - Wikipedia**

Other Properties: High Heat Capacity. It takes a lot of energy to heat up or cool

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down water. Animals are made of water, which helps them . regulate their body temperature.

## **Section 2.2 Properties of Water - Helena High School**

Miss Morrison Physical Science Section 2: Properties of Matter Page 21. Terms in this set (10) ... Physical Change. On a

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hot summer day, water vapor will condense into water droplets on the outside of a glass of iced tea. The glass of iced tea is a 2. \_\_\_\_\_ of sugar, tea, lemon, and water. 2. Mixture.

### **Section 2: Properties of Matter; Morrison Flashcards | Quizlet**

Section Summary. Water has many

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properties that are critical to maintaining life. It is polar, allowing for the formation of hydrogen bonds, which allow ions and other polar molecules to dissolve in water. Therefore, water is an excellent solvent.

## **2.2 Water - Concepts of Biology - 1st Canadian Edition**

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Chapter 2 Section 2 "Properties of Water". Use this activity to check your understanding of water, solutions, suspensions, acids, and bases. See a list of terms used in these activities.

## **Quia - Chapter 2 Section 2 "Properties of Water"**

Matter Section 2 Physical Properties,

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continued • Physical properties can help identify substances. • Physical properties can be observed or measured. – Examples: shape, color, odor, texture, state, melting point, boiling point, strength, hardness, magnetism, the ability to conduct electricity or heat

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## **Section 2: Properties of Matter**

The bill creates a commission to regulate, protect, conserve and control water resources identified in this bill as water sources crossing state boundaries in accordance with Section 2 as well as the first schedule of this act for equitable and sustainable social and economic development and to maintain

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environmental integrity.

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